CHIZHOV, A.F.

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PHASE I BOOK EXPLOITATION

SOV/4560

Tsentral'naya aerologicheskaya observatoriya

Trudy, vyp. 25 (Transactions of the Central Aerological Observatory, no. 25) Leningrad, Gidrometeoizdat, 1959. 83 p. 700 copies printed.

Sponsoring Agency: Glavnoye upravleniye gidrometeorologicheskoy sluzhby pri Sovete Ministrov SSSR.

Ed. (Title page): Ye. G. Shvidkovskiy; Ed.: Yu. V. Vlasova; Tech. Ed.: N. V. Volkov.

PURPOSE: This issue of the Transactions is intended for specialists in the physics of the atmosphere and aerology.

COVERAGE: This collection of 4 articles deals with problems connected with research of the upper atmosphere. The scientific use of artificial Earth satellites and rockets for the investigation of the upper layers of atmosphere is described. The energy distribution in the spectrum of solar rays in an

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Transactions of the Central Aerological (Cont.) SOV/4560

absolutely pure and dry air is calculated for the troposphere and for the lower part of stratosphere, taking into account absorption by ozone. Experimental data on the change of the temperature coefficient of resistance for tungsten wire under various conditions of preliminary heating, as well as data on the tensiometric coefficient of tungsten, used in rocket experiments are presented. The two articles by L. A. Biryukovs were written under the guidance of I. A. Khvostikov. References follow each article.

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Repnev, A. I. Properties of the Upper Atmosphere and Artificial Earth Satellites

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AVAILABLE: Library of Congress

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JA/dwm/ec 12-7-60

CHIZHOV, A.F.

Omegatron-type mass spectrometer. Trudy TSAO no.42:39-52 '62.

(MIRA 15:12)

(Mass spectrometry)

REPNEV, A.I.; CHIZHOV, A.F.

Some processes taking place in a mass spectrometer. Trudy TSAO no.42:53-74 162. (MIRA 15:12)

CHIZHOV, A.F.

Some preliminary results of developing the methodology for using a omegatron-type mass spectrometer in a quantitative determination of the composition of a gas mixture. Trudy TSAO no.42:75-83. 162. (MIRA 15:12)

(Gases-Analysis) (Mass spectrometry)

CHIZHOV, A.F.

Measuring the temperature of the free atmosphere with allowance for recombination of atoms. Trudy TSAO no.46:76-84 '63. (MIRA 17:1)

SOKOVA, N.A.; CHIZHOV, A.F.

Design of a laboratory setup for determining the atomic recombination coefficients. Trudy TSAO no.46:85-90 163. (MIRA 17:1)

VOLOSATOVA, A.I.; OZEROVA, A.S.; CHIZHOV, A.F.

Feeding and recording system for omegatron type mass spectrometers. Trudy TSAO no.46:101-105 '63. (MIRA 17:1)

L 23505-65 EWT(1)/EWT(m)/EPF(c)/SWG(w)/FCC/EEC-4/EPR/SEC(t)/EWP(t)/ GW--2 ACCESSION NR: ATL001568 S/2789/64/000/056/0000/001T AUTHOR: Sokova, N.A., Fedynskiv, A.V., Chizhov, A.F. TITLE: An investigation of the properties of the "omegatron" in measuring جوير جيده والإنجاب المناب SOLTHIE To serve the common agreens the skara observatoring. Trade of most constant in a second of the second of t and a second respect to the second let es, come a la la la la la la la la miliogen miliogen partine , fore-All a locks of the control of the integration mass or a tracket Action to the state of the partial pressure of molecular nitrogen in rarefied mixtures of atm solvers me attituinty with continuous acres or interspread Partnerspread Control Attended to the American Card 1/3

L 23505-65

ACCESSION NR: AT5001568

grams. This omegatron differs additionally from that described by Chizhov ITrudy TSAO, No. 42, page 39, 1962) in the use of non-magnetic nichrome for the electrodes and the use of an additional diaphragm for adjustment of the electron beam. Optimum values of the working preameters are: $H = 2 \cdot 10^3$ gauss, accelerating voltage of the ionizing electrons 140 volts, collector voltage 0.4 volts, amplitude of the high frequency field 0.3 volts, emission current 5 ma. Resolution for masses of the order of molecular nitrogen is 7.5. Determination of the relative proportions of neon 20 and neon 22 in a gaseous mixture by measuring the ion current of the device is accurate to + 10% to a partial neon pressure of 5.10⁻⁵ mm. Hg, which is no worse than measurements made using a model EMU-3 amplifying electrometer. In order to make absolute measurements, the device is calibrated by measuring the ion current as a function of introduced molecular nitrogen. In order to attain the desired stability of the current as a function of pressure, particular attention has to be paid to increasing electrical insulation (to prevent leakage losses), to improving the cleanliness of the electrode surfaces vo retard gaseous sorption effects) and to using a longer warm-up period before An evaluation of the distortion introduced by using the lan numer described by Kostko and Felynskiy (Trudy TSAO, No. 46, 1963) is pusic

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23505-65

ACCESSION NR: AT5001568

Comparative tables for the ion pump and oil and mercury diffusion pumps are given. It is concluded that sufficient accuracy is achieved with the ion pump for work with mass numbers of the order of molecular nitrogen. Orig. art. has: 8 figures and 2 tables.

ASSOCIATION: Tsentral naya aerologicheskaya observatoriya (Central aerologic observatory)

SUBMITTED: 00

ENCL: 00

SUB CODE: ES, GP

NO REF SOV: 005

OTHER: 001

Card 3/3

AUTHOF: Perov, S. F.; Chizhov, A. F.

TITLE: On the possibility of measuring the concentration of atomic oxygen at a height of 100-150 km by a heat-transducer method

SOURCE: Tsentral naya aerologicheskaya observatoriya. Trudy, so. 56

100-16214 Augusta and a stransferv. Teorems and a stransferv. Teorems and a stransferv. Teorems and a stransferv.

TOPIC TAGS: molecular omygen, atomic oxygen, heat transferve concentration of stemple and a stransferve at a stran

1 22612-65 ACCESSION MP: AT50/1373 heat, emission coefficient, and the Boltzmann constants there were Park Commence of the Commence Company of the second second snape of the instrument, and the dissipation energy of own An overheating process caused by the recombination takes place at an altitude of 90 km when the atomic concentration is 10" atoms per cubic centimeter. The maximum overheating under favorable conditions 2000. The main test of the measuring device is a system. wire. The trace il a smallbrium of the wire is express, gant to a solu in metweer recommon in the case, whose external heat, and the temperature of a measuring device out the out into a grass gox with the con-TO DESIGNATED IN THE DOX. HER. I A Compared which the concentration in the free atmosphere. The ... nast lifteers of teries, and 20 formulas. ASSOCIATION: Teentral naya aerologicheskaya observatoriya (Central Aerolegical Observatory) Cord 2/3

L:2612-65
ACCESSION NR: AT5C01573
SUJMITTED: 00 ENCL: 00 SUB CODE: ES
NÓ REF SOV: 012 OTHER: 006 ATD PRESS: 3172

CHIZHOV, A.F.

Methodology of measuring the density of chemically active components in application to Soviet radio-frequency mass spectrometers. Trudy TSAO no.56x128-131 *64 (MIRA 18x1)

SOKOVA, N.A.; CHIZHOV, A.F.

Use of the mass spectrometric method for determining partial densities and temperatures of components of the upper atmosphere. Trudy TSAO no.61: 28-43 '65. (MIRA 18:7)

CHIZHOV, A.F.

Use of the theory of free molecular current for processing data receiving by means of a quadrupole mass filter. Trudy TSAC with 18:7)

CHIZHOV, A.I.

Endocrine secretion of the pancreas during lowered gestric secretion. Terap. arkh. 30 no.7:66-71 J1 '58 (MIRA 11:8)

1. Iz fakul'tetskoy terapevticheskoy kliniki (dir. - prof. A.G. Gukasyan) sanitarno-gigiyenicheskogo fakul'teta Moskovskogo ordena Lenina meditsinskogo instituta imeni I.M. Sechenova. (GASTRIC JUICE.

secretion, relation to pancreatic juice secretion (Rus)) (PANCRMAS.

juice, secretion, relation on gastric juice secretion (Rus))

CHIZHOV, A. I.: Master Med Sci (diss) -- "The external-secretory function of the pancreas in hypo- and an-acid states of the stomach". Moscow, 1959. 8 pp (KL, No 12, 1959, 133)

HIZHOV, A.I., kand.med.nauk

Complications and treatment of patients in the immediate postoperative period following mitral commissurotomy. Terap.arkh. no.6:83-86 162. (MIRA 15:9)

1. Iz kardiologicheskogo otdeleniya (zav. - prof. V.Ye. Nezlin) Instituta serdechno-sosudistoy khirurgii (dir. - prof.S.A. Kolesnikov, nauchnyy rukovoditel akad. A.N. Bakulev AMN SSSR. (MITRAL VALVE-SURGERY) (POSTOPERATIVE CARE)

s/536/61/000/052/007/008 D201/D301

AUTHOR:

Chiznov, A.I., Engineer

TITLE:

Determining a gyromotor rotor starting time

SOURCE:

Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy, no. 52, 1961. Nekotoryye voprosy sovremennoy teknnologii

priborostroyeniya, 80 - 84

TEXT: The author considers results of partial automation of measurements of the starting time of the rotor of an industrial type gyrements of the starting time of the rotor of an industrial type gyro. Since during starting the gyro motor consumes a continuously ro. Since during starting the gyro motor consumes a continuously varying amount of current, the time dependence of starting current was determined. It was proved that a differentiating RC circuit could be used in conjunction with the stopper. Using such a circuit could be used in conjunction with the stopper. Using such a circuit act = TU in (T - the time constant of the differentiating network) and if Uin is a voltage across a constant resistance R in series with one of the gyromotor phases, the output voltage Uout of the RC circuit becomes proportional to Uin. Experimental graphs of start-

Determining a gyromotor rotor ...

S/536/61/000/052/007/008 D201/D301

ing current show that for values of $R \le 2$ ohms, the motor current is not affected and this feature allows automatic recording of the start time of the gyromotor. The automatic test arrangement has an electromagnet which starts a stopper when the power is switched on. The current taken produces a voltage drop across the constant resistor R, varying with the speed of the motor; it is initially +ve and then decreases to zero and becomes negative. When the -ve value of the voltage drop becomes 25 mV, a relay opens the circuit of the electromagnet which automatically stops the stopper. There are 5 figures.

Card 2/2



16.9500 (1031, 1121, 1132) 9.2530 S/103/61/022/002/012/015 B019/B060

AUTHORS:

Leskov, V. G., Chizhov, A. I., Chicherin, N. I. (Leningrad)

TITLE:

Some diagrams of half-wave (high speed) magnetic

amplifiers for servodrives

PERIODICAL:

Avtomatika i telemekhanika, v 22, no. 2, 1961, 250-258

TEXT: A study has been made of three diagrams of magnetic amplifiers displaying certain improvements compared with other known diagrams. The first part of the present paper is devoted to a discussion of a double-branch half-wave magnetic amplifier with a strong capacitive positive and feedback. The main elements of the circuit as well as its mode of operation are described with the aid of Fig. 1. If a magnetic amplifier of this kind has a phase-sensitive rectifier circuit as shown by Fig. 3 and as suggested by V. G. Baranovskiy, an output voltage will then be obtained owing to the properties of the magnetic amplifier, one component of which will be proportional to the input signal, while the second component will be proportional to the variation rate of the d-c component of the input voltage of the phase-sensitive rectifier. These properties Card 1/6

89181

Some diagrams of half-wave ...

S/103/61/022/002/012/015 B019/B060

indicate a good usability for servodrives. The second part deals with two double-branch half-wave magnetic amplifiers with a high Q factor. Magnetic amplifier controls are made by way of a change of the magnetic state of the cores through a change of the d-c component of the magnetic field. In magnetic amplifiers with a positive feedback this task can be solved either by changing magnitude and direction of the current in the control coils or by changing the positive feedback though a change of the rectification factor. The second possibility allows, as shown by tests, working out high-quality magnetic amplifiers. Fig. 4 shows a high-speed, push-pull magnetic power amplifier with a-c output. The respective control is done by changing the internal feedback with the aid of transistors controlling the feedback factor. This circuit has a large power amplification factor (larger than 1.5.105), low inertia. $(K_p/\tau \approx 7.5 \cdot 10^7, K_p$ being the power amplification factor, τ the time constant; moreover it is easy to assemble and has a large linear part of the characteristic. A further development of this diagram is shown in Fig. 5. As may be seen, this diagram dispenses with rectifiers in the

Card 2/6

89181

Some diagrams of half-wave ...

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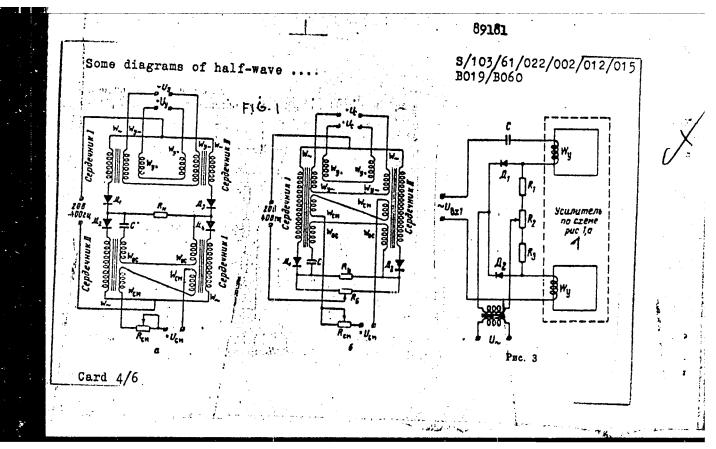
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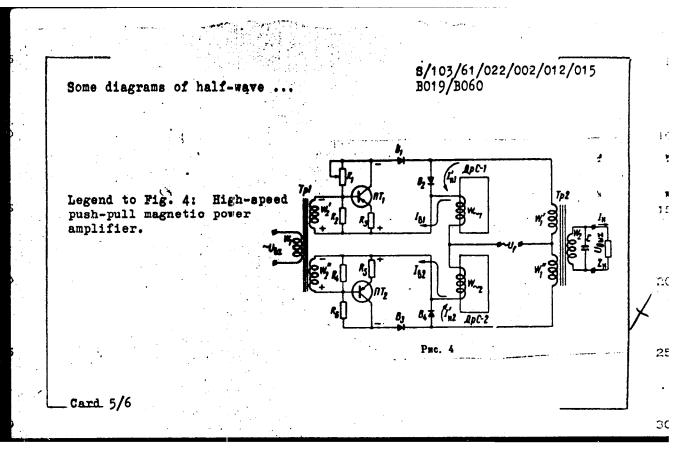
feedback circuit and replaces them by transistors. After a thorough discussion of the properties of this diagram it is pointed out that the joint use of transistors and magnetic amplifiers permits working out amplifier systems satisfying all requirements regarding operational safety, quick response, minimum weight, and size, provided the amplification factors and putput power are sufficiently large. There are 6 figures and 4 Soviet-

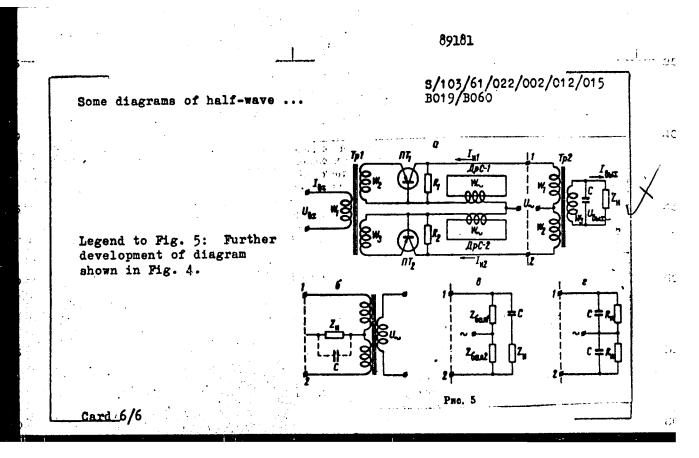
BUBLITTED: March 4, 1960

Legend to Fig. 1: Two variants of magnetic amplifiers.
Legend to Fig. 3: 1) amplifier according to diagram in Fig. 1a.
(See next card for figs.)

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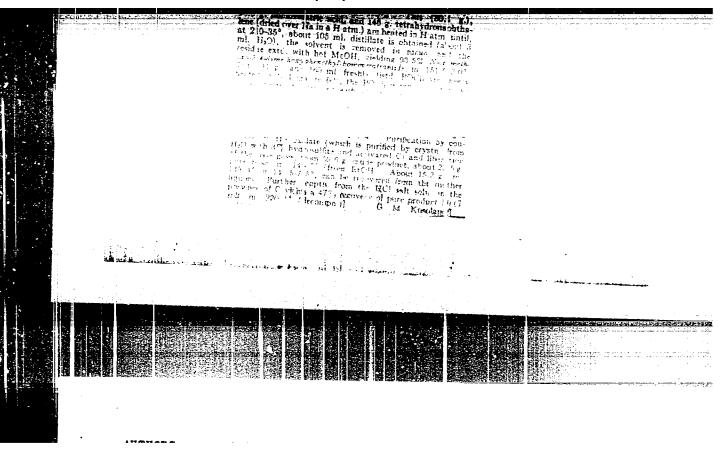


NAKHMANSON, V.M.; OSIDZE, D.F.; SEROV, M.F.; ALEKSANDROVA, V.T.; SOLOV'YEV, S.; MALYSHEV, N.; IVANENKO, N.M.; POTATURKIN, V.; CHIZHOV, A.I.; MIKHAYLOV, N.N.

In the Soviet Union. Veterinariia 39 no.1:88-96 Ja '63. (MIRA 16:6)

CHIZHOV, A.I. (Moskwa)

Potentials of mountain valleys. Priroda 53 no.4:123 '64. (MIRA 17:4)



79-28-5-10/69

Synthesis of Azubicyclo-(3,2,1)-Octane-7-Carboxylic Acid

The nicotine aldehyde (II) was condensed with malonic acid ester in the presence of piperidine at room temperature; there the diethyl ester of 2-(pyridil-3')-2-oxyethane dicarboxylic acid of the acid-1,1(II) formed. The heating of the latter (II) with acetic anhydride caused the splitting off of a molecule of water and the separation of diethylester of 2-(pyridil-3')--winyldicarboxylic acid-1,1 (III). The chlorine hydrate of (III) was reduced in alcohol solution in the presence of platinum oxide according to Adams, the chlorine hydrate of diethyl ester of the 2-(piperidil-3')-ethanedicarboxylic acid-1,1 (IV) having been obtained on this occasion. In order to further make possible the conversion from 3-substituted piperidine to the azubicyclic system compand(IV) was treated with bromine in chloroform. The synthetized dietnyl ester of 2-(piperidil-3')-1-bromethanedicarboxylic acid-1,1 (V) on heating with pyridine converted to 7,7 dicarboethoxy-1-azabicyclo-(3,2,1)-octane (VI). On boiling this diester with concentrated hydrochloric acid the chlorine hydrate of 1-azabicyclo-(3,2,1)-octane-7-carboxylic

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79-28-5-10/69

Synthesis of 1-Azubicyclo-(3,2,1)-Octane-7-Carboxylic Acid

acid (VII) resulted. There are 3 references, 1 of which is

Soviet.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevti-cheskiy institut (All-Union Scientific Chemical and Pharma-

ceutical Research Institute)

SUBMITTED: April 15, 1957

Card 3/3

AUTHORS:

Sergiyevskaya, S. I., Levshina, K. V., SOV/79-28-7-24/64

Chizhov, A. K., Gavrilova, A. I., Kravchenko, A. I.

TITLE:

N-Di(Ethyl Chloride) Amines of the Alicyclic Series. I(N-Di

(khloretil) aminy alitsiklicheskogo ryada. I)

PERIODICAL:

Zhurnal obshchey khimii, Vol 28, Nr 7,

pp. 1839 - 1845 (USSR)

ABSTRACT: The authors discuss the synthesis and some properties of the dichloroalkylamines of the cyclopentane-, cyclohexane- and cyclcheptanesseries. They synthetized the compounds of two -types: In the one (Formula I) the di(chlorealkyl) amino group is directly bound to the carbon of the nucleus, and in the other to the carbon of the side chain (II). The compounds of type (II) are alloyelic derivatives of methyl-N-bis (ethyl chloride) amine which is of importance for medicine. The two methods used most were employed for the synthesis of N-di(ethyl chloride) amine: according to the one [=(a) of Table 1] the ethylene oxide reacts with the amino compounds, according to the other [-(b) of Table 1] the compounds containing halogens are caused to react with diethanol amine. The final stage, i.e. the substitution of the hydroxyl groups by chlorine is the same

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M-Di(Ethyl Chloride) Amines of the Alicyclic Series. I SOV/79-28-7-24/64

for both methods, according to the specific characteristic features of the N-di(oxyethyl)amines. The synthesis of the dichloro-alkyl amines of type (I) had to be carried out according to method (a). The necessary alicyclic amines as initial products had been obtained in the cyclopentane- and cycloheptane series by the reduction of the ketone oximes, and in the cyclohexane series by the catalytic hydration of the aromatic amino compounds. The chloro-methyl derivatives of the same alicyclic hydrocarbons served as initial products for the synthesis of the compounds of type (II). The chloro-methyl cycloalkanes were obtained according to the reaction scheme mentioned. Thionyl chloride served as chlorination agent (I and II)(substitution of hydroxyl by chlorine). There are 2 tables and 8 references, 2 of which are Soviet.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S.Ordzhonikieze (All-Union Scientific Chemical and Pharmaceutical Institute imeni S. Ordzhonikidze)

Card 2/3

H-Di(Ethyl Chloride) Amines of the Alicyclic Series. I SOV/79-28-7-24/64

SUBMITTED: February 7, 1957

1. Dichloroalkylamines—Synthesis 2. Dichloroalkylamines—Properties 3. Cyclic compounds—Molecular structure 4. Ethyl chloride amines

Card 3/3

AUTHORS:

117/79-28-7-25/64 Sergiyevskaye, S. I., Levshins, E. V., Gavrilova, m. I.,

Chizhov, A. E.

TITLE:

N-Di (Chloro-Ethyl) amines .ith Alicyclic and Aromatic Radicals in the Molecules. II (N-di(khloretil)aminy a alitsikli-

cheskimi i aromatiche slimi redikalami v molekulukh. II)

FERIODICAL:

Zhurnal obshchey khimii, 1958, Vol. 28, Nr 7, pr. 1845-1849

(USSR)

ABSTRACT:

The sim of the present investigation was the synthesis of the a-mi(chloro-sthyl)chins; shien simultinequal; have an aromatic and an alicyclic resicul in the molecule. The structures of these compounds may be seen from the reaction scheme: the compounds (I) and (II) appear as arylated analogs of some H-di(enloro-ethyl)amines of the slicyclic ceries already earlier described by the authors (Ref 1). The compounds (III) differ from (I) and (II) by the not that the acomatic radical is not a component of the elicyclic ratical. The corresponds

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ing eyenogen compounds carved as initial products, viz , the nitriles alt- IN for the types (I) and (II), and the nitrate

N-9i (Chloro-Sthyl) Amines with Elicyclic and Aromatic Endicals in the acceptures. II

H-OH-CN. for type (III), where R denotes an alicyclic radical. $^{\rm C}6^{\rm H}5$

all these nitriler are assily obtained by the condensation of the cyanobenziles with 1,4-dibromobutane, 1,5-dibromopenture and bromocyclohexane in the presence of sodium amide. The reduction of the nitriles to primitry unines was carried out either catalytically with hydrogen or by means of lithium-aluminum hydride. The transition from amines to their N-di-(ethyloxy)-derivatives and from these to the N-di(enloroethyl amines took place according to reference 1. In the purification of the hydrogen chloride salts of the above mentioned amines the solvents had to be selected carefully. The nuthors synthesized the hitherto not described N-di(chloroethyl) amines and some other compounds of the cyclopentane- and cyclohexane series. There are 1 table and 5 references, 3 of which are Soviet.

Card 2/3

N-Di (Chloro-Ehyl) Amines With Alicyclic and Aromatic Radicals in the

ASSOCIATION:

Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordzhonikidze (All-Union Scientific Chemical and Pharmaceutical Research Institute imeni S. Ordzhonikidze)

SUBMITTED:

February 7, 1957

1. Ethyl chloride amines--Molecular structure 2. Ethyl chloride --Synthesis 3. Cyclic compounds--Chemical properties

Card 3/3

AUTHORS:

Chizhov, A. K.,

Rubtsov, M. V.

SOV/79-29-1-29/74

TITLE:

Synthesis of the 7-Monosubstituted Compounds of 1-Azabicyclo-[3,2,1]-Octane (Sintez 7-monozameshchennykh 1-azabitsiklo-[3,2,1]-oktanov)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 1, pp 130-136 (USSR)

ABSTRACT:

In the previous report (Ref 1) the authors described the synthesis of the 1-azabicyclo-[3,2,1]-octane-7-carboxylic acid. Owing to its reactive carboxyl group this acid can be used in the synthesis of various 7-monosubstituted compounds of the 1-azabicyclo-[3,2,1]-octanes. These compounds are very interesting in the biological research work as far as among the isomeric 2-monosubstituted compounds of quinuclidine (Ref 2) pharmacologically active products could be found. In the present paper the synthesis of the 7-monosubstituted compounds of 1-azabicyclo-[3,2,1]-octanes were described, i.e. of the amides, amines, hydrazides, esters, alcohols, halides, chloric acid anhydrides of the acids and some quaternary salts. For the synthesis of these compounds the 1-azabicyclo-[3,2,1]-octane-7-carboxylic acid (I) was used as initial product which was transformed into the chloric acid anhydride (II) and furthermore

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Synthesis of the 7-Monosubstituted Compounds of 1-Azabicyclo-[3,2,1] -Octane

SOV/79-29-1-29/74

into the 7-monosubstituted compounds of 1-azabicyclo-[3,2,1] -octanes according to the scheme mentioned. The synthesized
bases (VI), (XI), (XVIII), and (XX) were converted by methyl
iodide into the corresponding methiodides (VII), (XII), (XIX),
and (XXI). The dimethiodides of the 7-(1-diethyl-amino propyl)-1-azabicyclo-[3,2,1]-octane (XIX) and diethyl-amino ethyl
ester of the initial acid (XXI) show a pronounced blocking
action on the ganglia of the vegetative nerve system which as
regards its character approaches the effect of dioquine, the
dimethiodide of the diethyl-amino ethyl ester of the
quinuclidine-2-carboxylic acid (Ref 4). It was shown that the
halogen atom in the molecule of 7-chloro methyl-1-azabicyclo-[3,2,1]-octane, unlike the 2-chloro methyl quinuclidine
described in publications, has a high mobility and is able to
undergo condensation with the sodium malonic ester. There are
4 references, 3 of which are Soviet.

Card 2/3

"APPROVED FOR RELEASE: 06/12/2000 CIA-RDP86-00513R000308920006-3

Synthesis of the 7-Monosubstituted Compounds of 1-Azabicyclo- 3,2,1 -Octane

SOV/79-29-1-29/74

ASSOCIATION:

Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordshonikidze (All-Union Chemicopharmaceutical Scientific Research Institute imeni

S. Ordzhonikidze)

SUBMITTED:

November 14, 1957

Caard 3/3

5(3) AUTHORS:

Levshina, K. V., Chizhov, A. K., Sheynker, Yu. N., Sergiyevskaya, S. I. SOV/79-29-4-31/77

TITLE:

Sulfonic Esters of the Cyclohemane Diols and the 1,4-Butane Diol (Sul'fonovyye efiry tsiklogeksandiolov i 1,4-butandiola)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Hr 4, pp 1184-1188

(USSR)

ABSTRACT:

Some alkyl sulfonic esters of aliphatic diols proved to be useful active agents against some kinds of cancer. The authors had to decide whether the amount and structure of the radical of sulphur had any effect on the biological properties of the sulfonic esters of 1,4-butane diol, and whether the diol necessarily belonged to the aliphatic series. Alkyl sulfonic esters of 1,4-butane diol with the radicals C2H5,C3H7,cyclo-C6H11

and alkyl sulfonic esters of the isomeric cyclohexane diols (1,2;1,3;1,4) were synthesized. All these compounds were obtained through a transformation of the corresponding sulfochlorides with the diols in waterp-free benzene and in the presence of triethyl amine. The synthesis of the sulfochlorides was carried out according to references 2 and 3. The initial

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Sulfonic Esters of the Cyclohexane Diols and the 1.4-Butane Diol

SOV/79-29-4-31/77

cyclohexane diols were prepared by catalytic hydrogenation of the corresponding bivalent phenols. Particular interest was caused by the alkyl sulfonic esters of the 1,4-cyclohexane diol, since they are structurally closest to the highly active esters of 1,4-butane diol. For this reason not only methane, but also ethane and propane sulfonates were obtained. The mixture of stereo-isomeric 1,4-cyclohexane diols was separated by recrystallization into cis- and trans-compounds. Starting from the cis- and trans-diols two series of alkyl sulfonates were obtained. The assumed cis- and trans-forms, however, had identical melting points and physicochemical properties. Their infrared absorption spectra were identical as well, while those of the initial 1,4-diols and the isomeric alkyl sulfonates of the 1,2- and 1,3-cyclohexane diol were different (Figs 1,2). Thus it was proved that only one product forms by the reaction of the corresponding alkyl sulfochlorides with cis- and trans-1,4-cyclohexane diols, and that, consequently, ancinversion of the less stable form into the stabler one takes place in the course of the reaction. Since the change of the alkyl group in

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Sulfonic Esters of the Cyclohexane Diols and the 1,4-Butane Diol

SOV/79-29-4-31/77

the synthesized 1,4-alkyl sulfonates of the cyclohexane (methyl-, ethyl, propyl sulfonates) does not cause any sizable changes in the spectrum (Fig 3), it may be assumed that various alkyl sulfonates exhibit the very same configuration, and that the form in question is the stable trans-form. The biological properties of the compounds obtained generally correspond to those of "milerane" (Mileran). There are 3 figures, 1 table, and 8 references, 3 of which are Soviet.

ASSOCIATION:

Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordzhonikidze (All-Union Scientific Chemopharmaceutical Research Institute imeni S. Ordzhonikidze)

SUBMITTED:

February 10, 1958

Card 3/3

CHIZHOV, A.K.; LEVSHINA, K.V.; SERGIYEVSKAYA, S.I.

Bis (-chloroethyl) aminomethyl derivatives of azobenzene. Fart
1: Method of synthesizing bis (-chloroethyl) amines of 4-substituted4'-methylazobenzene. Zhur. ob. khim. 30 no.11:3695-3700 N'60.

1. Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordshonikidse.

(Azobenzene)

"APPROVED FOR RELEASE: 06/12/2000 CIA-RDP86-00513R000308920006-3

CHIZHOV, A.K.; LEVSHINA, K.V.; SERGIYEVSKAYA, S.I.

Bis (B-chloroethyl) amines of bicyclic compounds. Part 3:

Some derivatives of bensocycloheptane with substituents in position 7 of the bicyclic compound. Zhur. ob. khim. 30 no.11:3700-3702 N'60.

(MIRA 13:11)

1. Vsesoyusnyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordshorikidse.
(Cycloheptabenzene)

3/080/60/033/04/44/045

AUTHOR:

Chizhov, A.K.

TITLE:

The Production of 6-Cyano-Benzocycloheptane-Imine-7

PERIODICAL:

Zhurnal prikladnov khimii, 1960, Vol 33, Nr 4, pp 988 - 989

TEXT:

A special device was developed for the reaction. It is shown in a diagram. Into a reaction retort with a capacity of 2,000 ml, 500 ml of absolute ether is poured and a lithium wire of 5.6 g and 1.5 mm in diameter is put into it. During mixing in a nitrogen flow 64 g of brombenzene is added. The temperature of the reaction mixture is maintained at 40°C. After mixing for 30 minutes 100 g of monomethylaniline is added in the course of 1 hour. Then in the course of 20.5 hours, 300 ml of a benzene-ether solution of 41 g of o-di-(\$\beta\$-cyanoethyl)-benzene is added. The mixture is cooled to room temperature and 300 ml of water is added. It is heated at 50°C for 1 hour and 5 minutes, it is cooled, a precipitate is filtered off. The ether-benzene layer is separated from the water layer and is dried with potash. A precipitate is formed which is dissolved by heating the retort on a water bath. The warm solution is filtered and evaporated in the vacuum at 65 - 70°C. The crystallizing reaction product 1s filtered off after standing for 12 hours and washed on the filter with ether. 25.7 g

Card 1/2

CHIZHOV, A.K.; LEVSHINA, K.V.; SERGIYEVSKAYA, S.I.

Bis (β -chloroethyl) aminomethylazobensenes and some analogous compounds. Zhur. ob. khim. 31 no.4:1288-1297 Ap '61.

(MIRA 14:4)

1. Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imoni S. Ordzhonikidze.

(Azobenzene)

CHIZHOV, A.K.

Derivatives of 1-asabiscyclo-(3, 2, 1)-octane, analogous to quinine alkaloids. Zhur.ob.khim. 31 no.10:3469-3477 0 '61.

(MIRA 14:10)

1. Vsesoyusnyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S.Ordshonikidze. (Quinine) (Alkaloids)

CHIZHOV, A.K.; LEVSHINA, K.V.; SERGIYEVSKAYA, S.I.

Bis (3-chloroethyl) aminomethyl derivatives of azobenzene. Part 3: p-Hydrozy-m-(o)-carboxy-p'-bis (3-chloroethyl)-aminomethylazobenzenes, their derivatives and analog compcunds. Zhur. ob. knim. 34 no. 5:1587-1592 My '64. (MIRA 17:7)

1. Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S.Ordzhonikidze.

"APPROVED FOR RELEASE: 06/12/2000 CIA-RDP86-00513R000308920006-3

L 00464-66 EVT(1)/FCC/E/A(h) GW

ACCESSION NR: AT5013406

UR/2789/65/000/061/0028/0043

40 40

AUTHOR: Sokova, N. A.; Chizhov, A. F.

4× ≠+1

TITLE: Use of the mass spectrometric technique for determining partial densities and temperatures of the components of the upper atmosphere

SOURCE: Tsentral naya aerologicheskaya observatoriya. Trudy, no. 61, 1965. Fizika vysokikh sloyev atmosfery, teoriya i metody issledovaniya (Physics of high atmospheric layers, theory and methods of investigation), 28-43

TOPIC TAGS: mass spectrometer, upper atmosphere, upper atmosphere density, upper atmosphere temperature

ABSTRACT: Analytical expressions were obtained which make it possible to calculate the absolute and relative partial densities of the components of atmospheric air both in the laboratory and in the free atmosphere from primary mass spectra. The expressions obtained can be used for processing mass spectra of atmospheric air. It is shown that many expressions are considerably simplified in various high-altitude intervals. A method of calculating the temperature of the free atmosphere for the case of multiple lines of the mass spectrum is proposed. It is shown that the omegatron is the most sensitive mass spectro-

"APPROVED FOR RELEASE: 06/12/2000 CIA-RDP86-00513R000308920006-3

L 00464-66

ACCESSION NR: AT5013406

meter for measuring the temperature of the free atmosphere. The need for a more accurate experimental determination of the sensitivity coefficients S_{ij} and S'_{ij} is emphasized. Orig. art. has: 2 figures, 2 tables, and 51 formulas.

ASSOCIATION: Tsentral neya aerologicheskaya observatoriya (Central Aerological

Observatory)

44,55

SUBMITTED: 00

ENCL: 00

SUB CODE: ES,GP

NO REF SOV: 032

OTHER: 009

Card 2/2

00467-66 EWT(1)/FCC/EWA(h) GW

ACCESSION NR: AT5013407

UR/2789/65/000/061/0044/0049

AUTHOR: Chishov, A. F.

TITIE: Use of the theory of free molecular flow for processing data obtained with a quadrupole mass filter

SOURCE: Tsentral naya merologicheskaya observatoriya. Trudy, no. 61, 1965. Fizika vysokikh sloyev atmosfery, teoriya i metody issledovaniya (Physics of high atmospheric layers, theory and methods of investigation), 44-49

TOPIC TAGS: mass spectrometer, upper atmosphere composition; ion source, quadrupole mass filter, atmospheric data processing

ABSTRACT: In mass-spectrometric measurements of the partial densities of the components of the upper atmosphere, one of the main difficulties is the loss of atoms on the walls of the measuring instrument. It is desirable, therefore, to construct a mass spectrometer in which this loss is reduced to a minimum. The article considers one of the possible theoretical schemes of interaction between the oncoming flux and the ion source of the mass spectrometer used by E. I. Schaefer (Journ. of Geoph. Res., Vol. 68, No. 4, 1963), in which use is made of Cord 1/2

I. 00467-66

ACCESSION NR: AT5013407

lesser degree than in the existing radio-frequency mass spectrometers, but the Schaefer ion source disturbes the oncoming flux considerably. The mass selection coefficient of the Schaefer source indicates that the latter is not free of mass selection. Partial densities of the main components of the free atmosphere, obtained by recalculating the experimental data of Schaefer (E. I. Schaefer, M. N. Nicholes, The University of Michigan. Ann Arbor, Mich., USA, NASA Contract No. NASr - 54 (05), Washington, D.C., USA) in accordance with the proposed theory, basically agree with the results obtained by other authors. Orig. art. has: 6 figures and 5 formulas.

ASSOCIATION: Tsentral'naya aerologicheskaya observatoriya (Central Aerological

SUMMITTED: DO

ENCL: 00

SUB CODE: ES, DP

NO REF SOV: 017

OTHER: 003

Cord 2/2

CHIZHOV, A.M., kand.tekhn.nauk

Experimental check on the stability of reinforced concrete beams. Bet.i zhel.-bet. 9 no.12:545-547 D 163. (MIRA 17:2)

CHIZHOV, A.M., inzh.; SIVERTSEV, I.N., doktor tekhn.nauk, prof., otv.red.

talculations for the reinforced concrete construction of ships for tensile and flexural strength] Raschet sudovykh zhelezobetonnykh konstruktsii na izgib s rastiazheniem. Gor'kii, GIIVT, 1959. 29 p (Gorkiy Institut inzhenerov vodnogo transporta. Trudy, no. 26)

(Naval architecture—Tables, calculations, etc.)

(Ships, Concrete)

CHIZHOV, A.M., inzh.

Using precast reinforced concrete for stationary ships. Rech. transp. 18 no.1:26 Ja 159. (MIRA 12:2) (Ships, Concrete)

YEGOROV, N.M., kand.tekhn.nauk; CHIZHOV, A.M., insh.

Erroneous recommendation. Rech.transp. 18 no.9:25-26
(MIRA 13:2)

(Ships, Concrete)

"APPROVED FOR RELEASE: 06/12/2000 CIA-RDP86-00513R000308920006-3

CEIZEOV, A.M., Cand Tech Sci - - (diss) "Consideration of Reinforced concrete ship construction on bending with tension,"

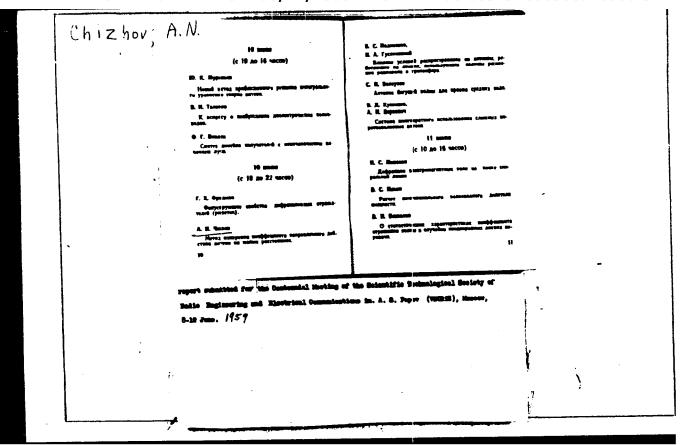
Gor'kly, 1960, 20 pp (Gor'kly Institute of Engineers of Water Transport) (aL, 34-60, 123)

CHIZHOV, A.M., kand.tekhn.nauk

Ways of improving the construction of section joints in reinforced concrete ships. Sudostroenie 29 no.5:54-56 My '63. (MIRA 16:9) (Ships, Concrete)

CHIZHOV, A.N. Studying sludge ice phenomena in mountain rivers. Trudy 00I no.65:84-99 158. (NIRA 12:1)

"APPROVED FOR RELEASE: 06/12/2000 CIA-RDP86-00513R000308920006-3

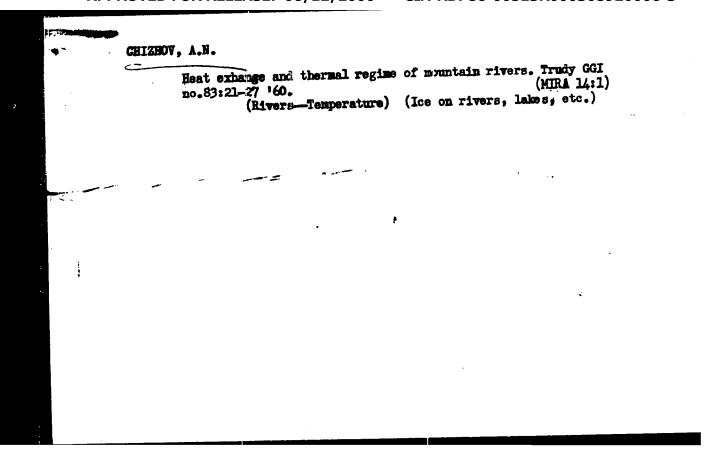


"APPROVED FOR RELEASE: 06/12/2000 CIA-RDP86-00513R000308920006-3

List on days residency and the formal position processes (to indicate the following states and t
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CHIZHOV, A.N.

Method of studying frazil ice in mountain rivers. Trudy GGI no.90: 103-122 '60. (MIRA 14:1) (Ice on rivers, lakes, etc.)



Formation and flow of frazil ice in mountain rivers. Trudy GGI no.93:3-23 '62." (MIRA 16:3) (Naryn RiverIce on rivers, lakes, etc.)					
	t.				

CHIZHOV, A.N.

Some types of mountain stream ice regimes. Trudy GGI no.103: 57-69 '63. (NIRA 16:7)

"APPROVED FOR RELEASE: 06/12/2000 CIA-RDP86-00513R000308920006-3

CHIZHOV, A.N.

Formation of ice jums on mountain rivers. Trudy GGI no. 129:44-66 165. (MIRA 18:10)

CHEPIGIN, G. V., insh.; NEKHAY, S. M., insh.; GUL', N. S., insh.; CHIZHOV, A. P., insh.

Replacing the double-cleaning cil filter with a full-flow centrifuge. Mashinostroenie no.5:95 S-0 162.

(MIRA 16:1)

(Tractors-Engines-1011 filters)

CHEPIGIN, G.V., kand.tekhn.nauk; GUL; N.S., inzh.; CHIZHOV, A.P., inzh. KHESIN, A.Ya.

Results of the operational tests of a full-flow RMTs device on the SMD diesel engine. Trakt. i sel'khozmash. 32 no.6:12-14 Je '62. (MIRA 15:6)

1. Dnepropetrovskiy sel*skokhozyaystvennyy institut (for Chepigin, Gul*, Chizhov). 2. Gosudarstvennoye spetsial*noye konstruktorskoye byuro po dvigatelyam (for Khesin).

(Tractors—Oil filters)

CHEPIGIN, G.V., kard. tekhn. nauk; GUL, N.S., inzh.; CHIZHOV, A.P., inzh.

Use of cast-iron crankshafts in motor-vehicle and tractor engines.
Meshinostroenie no.5:112-113 S-0 '63. (MIRA 16:12)

1. Dnepropetrovskiy sel'skokhozyaystvennyy institut.

CHEPIGIN, G.V.; GUL', N.S.; CHIZHOV, A.P.

Experiments in the use of cast iron crankshafts in tractor diesel engines. Trakt. i sel'khozmash. 33 no.8:44-45 Ag '63. (MIRA 16:11)

1. Dnepropetrovskiy sel'skokhozyaystvennyy institut.

CHIZHOV, A.T., kandidat tekhnicheskikh nauk.

Mechanization of ballast work. Transp.stroi. 6 no.2:27-28 F 156. (Ballast) (MIRA 9:6)

CHIZHOV, A.T. HOVOZHILOV, G.I., kand. tekhn. nauk; CHIZHOV, A.T., kand. tekhn. nauk.

New working methods in assembling yards. Transp. stroi. 7 no.11: 27-28 N 157. (MIRA 11:2) (Loading and unloading) (Railroads--Construction)

CHIZHOV, A.T

CHIZHOV, A.T., kand.tekhn.nauk

"Mechanization of track maintenance on foreign railroads."

V.S.Besruchko, V.I. Platov, K.E. Ivanov. Reviewed by A.T.Chishov.

Put' i put.khoz. no.12:47 D '57. (MIRA 10:12)

(Railroads--Maintenance and repair) (Besruchko, V.S.)

(Platov, V.I.) (Ivanov, K.E.)

CHIZHOV A.T. kand telhn.nauk

Economic and technical indexes of the rolling of earthwork using trains. Transp.stroi. 9 no.9:50-51 S '59. (MIRA 13:2)

(Railroads--Track)

CHIZHOV, A.T., hand. tekhn. nauk

Small-scale mechanisation in maintaining railroads in the German Democratic Republic. Transp. stroi. 9 no.11:54 H '59 (MIRA 13:3) (Germany, Bast--Railroads--Equipment and supplies)

CHIZHOV, A.T., kand.tekhn.nauk

Marking the 150th anniversary of the Leningrad Railroad
Engineers Institute. Transp.stroi. 9 no.12:54 D 159.

(MIRA 13:5)

(Leningrad--Railroad engineering--Study and teaching)

CHIZHOV, A.T., kand.tekhn.nauk; KOSTYUKOVICH, A.R., insh.

New developments in laying switches. Transp. stroi. 12 no.8: 14-16 Ag '62. (MIRA 15:9)

ROZANOV, A.A., inzh.; CHIZHOV. A.T., inzh.; ULANTSEV, I.D., inzh.

Manual on the mechanization of ballasting. Transp.stroi. 13 no.9:
73-74 S '63. (MIRA 16:12)

CHIZHOV, A.T., kand.tekhn.nauk

Transporting railroad switches on a rollingstock. Transpostroi.
13 no.9:7-8 S *63. (MIRA 16:12)

CHIZHOV, A.T., kand.tekhn.nauk

Replacing one-piece timbers. Transp. stroi. 12 no.3:12-14 Mr 162. (MIRA 16:11)

ZAYTSEV, P.I.; LIZOGUB, I.G.; PETRUKOVICH, A.A., zasl. deyatel' nauki i tekhniki Uz.SSR; SMYKOV, Ye.K.; CHIZHOV, A.T.; YAKOBSON, S.I.; ANDREYEV, G.Y., dots., retsenzent; CRECHUK, V.S., dots., retsenzent; NEKHAY, V.T., red.

[Mechanization of the assembly, laying and exchange of switches] Mekhanizatsiia sborki, ukladki i smeny strelcchnykh perevodov. Minsk, Vysshaia shkola, 1964. 69 p. (MIRA 18:3)

1. Leningradskiy institut inzhenerov zheleznodorozhnogo transporta, kafedra "Zheleznodorozhnyy put" (for Andreyev, Grechuk).

ACCESSION NR: AP4014676

5/0108/64/019/001/0057/0062

AUTHOR: Chizhov, A. V.; Dobzhinskiy, B. N.

TITLE: Errors in measuring variable frequency by the counting-zeros method

SOURCE: Radiotekhnika, v. 19, no. 1, 1964, 57-62

TOPIC TAGS: FM variable frequency, FM variable frequency measurement, counting zeros frequency measurement, frequency measurement error

ABSTRACT: The counting-zeros method for the accurate measurement of "instantaneous" frequency in FM circuits is theoretically considered. Two types of error are inherent in the method: (1) the error due to the nonlinear nature of frequency variation and (2) the error due to the discrete-counting device. The relation between the values of these errors and the averaging time of measurement is investigated. This formula for selecting the averaging time is recommended:

Card 1/2

ACCESSION NR: AP4014676

$$\xi(\Delta_{1,\text{max}} T_{\text{a}}, t_{\text{mex}}) > T_{\text{cl}} > \frac{2}{\Delta_{1,\text{max}}}$$

where Δ_1 and Δ_2 are errors due to the two causes mentioned above, T_m is the FM period, t_{max} is a moment of time corresponding to Δ_1 maximum, T_{α} is the averaging time, and ξ is a function $T_{\alpha} = \xi(t_{max} \mid \Delta_1, t_{max} \mid \lambda_2, t_{max} \mid \Delta_3, t_{max} \mid \Delta_4, t_{max} \mid \Delta_4, t_{max} \mid \Delta_5, t_{max}$

ASSOCIATION: none

SUBMITTED: 02Mar62

DATE ACQ: 07Feb64

ENCL: 00

SUB CODE: GE

NO REF SOV: 000

OTHER: 001

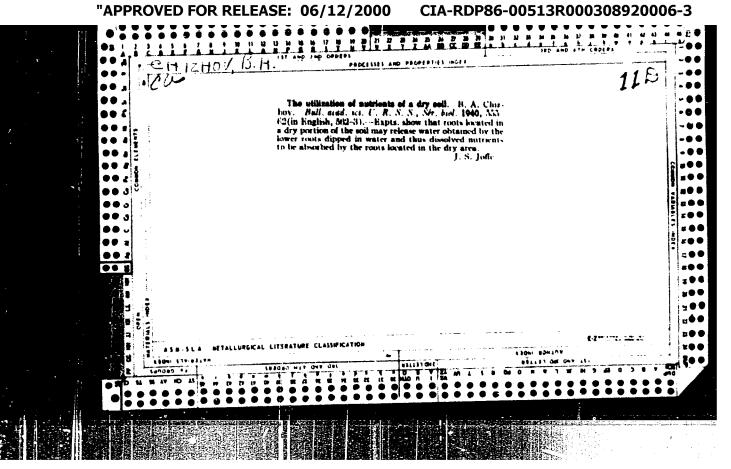
VDOVENKO, V.M.; KRIVOKHATSKIY, A.S.; CHIZHOV, A.V.

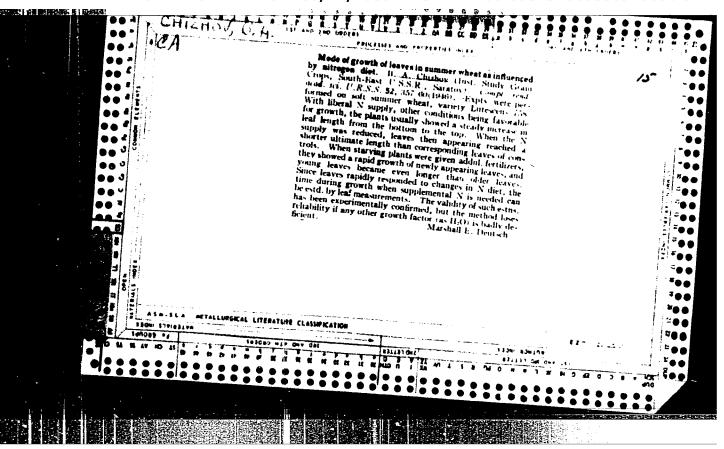
Extraction of chlorides with mixed solvents. Zhur. neorg. khim. 5 no.10:2363-2365 0 '60. (MIRA 13:10) (Chlorides)

PARAMONOVA, V.I.; KEREYCHUK, A.S.; CHIZHOV, A.V.

Ion exchange in the study of the state of a substance in solution. Part 9: Complex formation of europium with some dibasic acids. Radiokhimiia 5 no.1:63-73 163.

(Europium compounds) (Acids, Organic) (Ion exchange)





CHIZHOV, B. A.

"Fertilizing the Field Crops in Grassland Crop Rotations in the coutheastern Part of the USSR," Saratov Oblast' gos. izd-vo, 1949

CHIZHOV, B. A.

30392

Opryedyelniye foz rayevitiya zachatochnogo kolosa u yarovoy pshyenitsy po poyavlyeniyu list'yev. Trudy In-ta Fiziologii rastyeniy im. timiryazyeva, T. VI, vyp. 2, 1949, S. 282-90.—Bibliogr: 9 Hazv.

SO: Letopis' No. 34

CHIZHOV, B.A.

"Composition, Properties, and Use of Fertilizers," Saratov Chlast' gos. izd-vo, 1950

CHIZHOV, B. A.

"Local Granular Fertilizers," Saratov Oblast' gos. izd-vc, 1951

CHIZHOV, B. A.

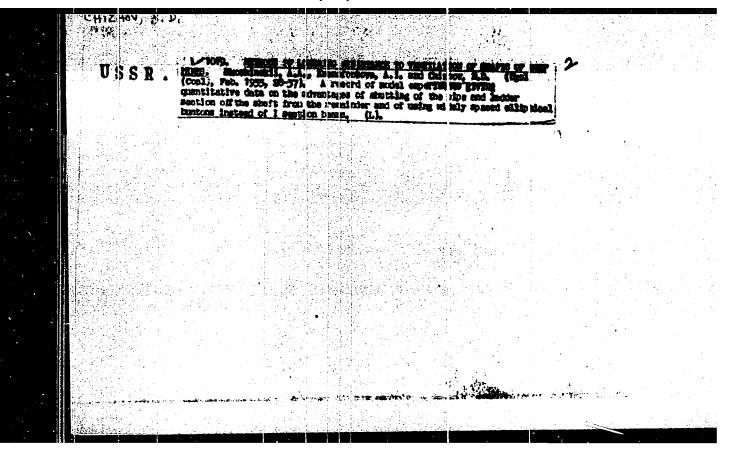
"Peruliarities in the Use of Fertilizers in the Cultivation of Crops on Irrigated Lands Beyond the Volga," Sov. agron., 10, No.2, 1952

CHIZHOV, E. D.

CHIZHOV, E. D. -- "Downflow Ventilation of Cleaning Places in Some Deep Level Gas Mines of the Stalino-Makeyevka Rayon Working Plat Sloping Coal Seams." *(Dissertations for Degrees in Science and Engineering Defended at USSE Higher Educational Institutions)
Min Higher Education USSE, Moscow Mining Inst iment I. V. Stalin, Moscow, 1955.

SO: Knizhnaya Letopis! No. 31, 30 July 1955.

*For the Degree of Candidate in Technical Sciences.



VOROPAYEV, Aleksandr Frolovich; KREMNEV, O.A., doktor tekhn. nauk, retsenzent; CHXZHOV, B.D., otv. red.; RATNIKOVA, A.P., red. izd-va; SHKLYAR, S.Ya., tekhn. red.

[Temperature control in deep mines] Upravlenie teplowym rezhimom v glubokhikh shakhtakh. Moskva, Gos. nauchmo-tekhm. izd-vo lit-ry po gornomi delu, 1961. 246 p. (MIRA 15:2) (Mine ventilation) (Heat—Transmission)

BARATOV, E.I., kand.tekhn.nauk; CHERNYAK, V.P., inzh.; CHIZHOV, B.D., kand.tekhn.nauk

Determining the rated parameters of outside air for purposes of predicting and regulating the heat conditions of mines located in areas with an extreme continental climate. Trudy Sem.po gor.teplotekh. no.4:72-74 *62. (MIRA 15:8)

1. Institut teploenergetiki AN UkrSSR.
(Mine ventilation)